

## ***The Continuous Culture of Microorganisms:***

### ***Continuous Culture System***

- *A microbial population of can be maintained in the exponential growth phase and at a constant biomass concentration for extended periods.*
- ***Open System:*** *system with constant environmental conditions maintained through continual provision of nutrients and removal of wastes .*

### ***Two Common Major Types of Continuous Culture Systems:***

#### ***(1). Chemostats***

- *Sterile medium is fed into the culture vessel at the same rate as the media containing the Mos is removed.*

#### ***(2). Turbidostats***

- *Photocell that measures the absorbance or turbidity of the culture in the growth vessel.*

## ***Environmental Factors on Growth:***

### ***The growth of Mos are effected by Chemical and Physical surroundings:***

- *Procaryotes are present anywhere life can exist.*
- ***Extremophiles:*** *Mos grow in harsh environments Live 1.5 miles below the earth's surface, w/o oxygen, and below 60°C.*

## ***Temperature***

- *Microbial cell temperature directly reflects that of the cell's surrounding.*
- *Most bacteria can grow over a **temperature range** of about 30° or more but have a narrow range for **optimal growth**.*
- *As we decrease the temperature below the optimum, we see a decline in growth rate that is consistent with enzymatic activity, but then it becomes very steep, giving rise to a fairly well defined minimal growth temperature.*
- *Above the optimum temperature, we see the growth rate decline very steeply, which gives rise to a sharply defined maximum growth temperature*

*It is not known what sets the upper and lower temperature although they are thought to*

- *reflect properties of the membrane lipids,*
- *effects on protein conformation, and/or initiation of protein synthesis.*

***Temperature Sensitivity of Enzyme-Catalyzed Reactions:***

- *A temp rise, increases the growth rate due to the velocity of an enzyme-catalyzed reaction.*
- *Velocity will double for every 10° C rise in temperature.*
- *As rate increase, the metabolism is more active at higher temp, Mo grow faster.*
- *Example: 10 -- 30, Velocity is 15 What is the velocity of the cell at 50° C?*

***High Temperatures:***

- *Damage MOs by denaturing enzymes, transport carriers, and other proteins*
- *Membranes are disrupted, lipid bilayer simply melts and disintegrates.*

***Low Temperatures:***

- *membranes solidify and enzymes don't work properly.*

***The temperature range of an organism can be used as a classifying characteristic. All bacteria have distinct cardinal temperatures:***

***Cardinal Temperatures Growth Temperatures:***

- *Minimum*
- *Optimum*
- *Maximum*

*Pc can grow at much higher Temp than EC*

***Psychrophiles*** *can grow at temperatures between 0-20°, optima growth is 15°*

- *Frequently found in naturally cold waters and soils.*  
*Such as the Artic and Antarctic.*
- *Examples include the Pseudomonads and Bacillus,*
- *Enzymes, transport systems and protein synthetic mechanisms function well as low temp.*

- Cell membrane have high levels of unsaturated fatty acid and remain semifluid when cold.

**Psychrotrophs or Facultative Psychrophiles** can grow at 0 to 7 ° C.

- Optima 20-30 ° C
- Maxima 35 ° C
- Psychrotrophic bacteria and fungi are important in spoilage of refrigerated foods

Most bacteria are **Mesophiles** and grow between 20-45 °C.

- Those that are found in the mammalian body have an optimum temperature of 37-44 ° C, Maxima is 45 ° C.
- Those found in the environment have an optimum of about 30 degrees C
- Almost all human pathogens are mesophiles, env. is around 37°C.

**Thermophiles** grow at temperature of 55° C or higher.

Minimum of 45 °C and optima between 55 and 65 ° C.

- Majority of prokaryotes
- Flourish in composts, self-heating hay stacks, hot water lines, and hot springs.
- More heat stable enzymes and protein synthesis systems / funct at higher temp.
- Membranes lipids more saturated and have higher melting points causing membrane to remain in tact at higher temp.
- These organisms are extremely useful in that they serve as sources for exceptionally stable forms of enzymes (i.e. bacillus stearothermophilus)

**Hyperthermophiles** are thermophiles that can grow at 90° C or above,

- Prokaryotes growth optima between 80 and 113 ° C.
- Do not grow well below 55 ° C.
- These organisms are extremely useful in that they serve as sources for exceptionally stable forms of enzymes (i.e. bacillus stearothermophilus)

### **pH (Acidity or alkalinity)**

pH is a measure of the Hydrogen Ion activity of a solution and is defined as the negative logarithm of the hydrogen ion concentration (expressed in terms of molarity).

pH scale 0.0 ( $1.0 \text{ M H}^+$ ) – 14.0 ( $1.0 \times 10^{-14} \text{ M H}^+$ )

Each unit represents a tenfold change in hydrogen ion concentration.

- **Bacteria can also be classified by the pH ranges in which they grow.**
- The internal pH of the cell remains close to neutral,
- An organism's tolerance to fluctuations in pH reflects the capacity of the membrane pumps to maintain that pH

**Acidophiles** *grow best below pH 4.0.*

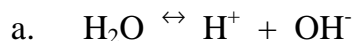
- The vinegar forming acetobacter and some of the sulfur oxidizing bacteria can tolerate the pH values as low as ~0 (the pH of 1N sulfuric acid)
- Growth range between 0 – 5.5

Neutrophiles

- Growth range is 5.5 to 8.0
- **Most bacteria and protozoa**

**Alkalophiles** (most grow best above ~pH 10)

- Growth range 8.5 to 11.5
- a few of the urea splitters, *Alcaligenes faecalis*, and *vibrio cholerae* can **thrive at pH levels as high as 9 and can tolerate levels greater than 10.**
- *E. coli* cannot withstand pH conditions greater than 8 or below 4.5.



$[\text{H}^+] = [\text{OH}^-] = 1.00 \times 10^{-7}$  mole per liter in pure water.

b.  $\text{pH} = \log 1/[\text{H}^+]$

The pH of pure water is  $\log (1)/(1.00 \times 10^{-7})$   
 $= \log (1.00 \times 10^7) = 7.00.$

c. Acids are pH 0 - 7, bases are pH 7 - 14.

d. Every increase of 1 pH unit is a 10-fold decrease in  $[\text{H}^+]$ .

*Most bacteria grow between pH 5 and pH 8 (across a 1,000-fold difference in external hydrogen ion concentration).*

A few extreme acidophiles (oddly, all are eukaryotes and archaea) can grow at or near pH 0.

## ***Oxygen Concentration***

### ***Oxygen effects on growth***

- a. O<sub>2</sub> reacts with certain enzymes in the cell to form hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and superoxide (O<sub>2</sub><sup>-</sup>). These compounds can damage biological macromolecules.
- b. Detoxifying enzymes: catalase, peroxidase, and superoxide dismutase.
- c. Singlet oxygen (esp. produced in photosynthesis) can be quenched by carotenoids
- d. Categories of microorganisms (by oxygen environment):
  - (1) Strict aerobes:
    - Organism grow in the presence of atmospheric O<sub>2</sub>
  - (2) Strict anaerobes
    - Do not tolerate O<sub>2</sub> at all and die in the presence of it.
  - (3) Facultative anaerobes
    - Do not require O<sub>2</sub> for growth but do grow better in its presence.
  - (4) Aerotolerant anaerobes
    - Grow equally well whether it is present or not
  - (5) Microaerophiles (low oxygen)
    - damaged by the normal atmospheric level of O<sub>2</sub> (20%) and require O<sub>2</sub> levels below the range of 2 to 10% for growth.

*Most organisms are very sensitive to the salinity of the environment or osmotic pressure.*

*The rigid structure of the bacterial cell wall enables it to grow over a wide range of osmotic pressures.*

*Most bacteria grow in ranges between 0.85% NaCl (physiological saline) and 3.5% NaCl (seawater)*

*Some bacteria have adapted mechanisms which enable them to withstand extremes of osmotic pressure*

*Halophiles grow best in environments where the osmotic pressure ranges between 18-24% NaCl.*

***Salinity (osmotic balance):***

- [hi H<sub>2</sub>O in/ low H<sub>2</sub>O out (hi salt) =>becomes dehydrated;
- low salt environment, H<sub>2</sub>O rushes in and cell lyses
  
- Many microbial environments are salty. For example, the oceans cover about 71% of the Earth's surface. Cells must maintain an osmotic balance with the environment while not allowing high salt concentrations to inhibit essential metabolic processes.
  
- Salt concentration in human blood is about 0.85% NaCl (physiological saline).
  
- Sea water contains about 3.5% salt (mostly NaCl). The Dead Sea has a salt concentration of about 24%.
  
- A saturated salt solution contains about 35% NaCl.

- Microorganisms that grow in extreme high-salt environments (15-30% NaCl; salt pans, evaporating pools, etc.) are called halophiles.

halophile vs. halotolerant

- d. Compatible solutes are produced to increase solute levels within cell without inhibiting cellular processes allowing the
- e. Bacteria in the genus Halobacterium grow best in environments where the NaCl concentration is above 18%. They actively pump Na<sup>+</sup> ions out of the cell and pump K<sup>+</sup> ions in to maintain their osmotic balance. Energy for pumping is generated by a unique light-dependent photosynthetic process that is not based on chlorophyll.

### ***Growth of bacterial cell cultures (populations)***

1. Doubling time or generation time: the time between one cell division and the next (the time it takes for the population of cells to double in size).
  - a. Doubling time depends upon the species and the culture conditions.
  - b. Optimal doubling times: *E. coli* (20 minutes) versus *Mycobacterium tuberculosis* (15-16 hours).
  - c. Natural environments: *E. coli* in the human intestine (10 hours), *Pseudomonas aeruginosa* in soil (2-3 days).

## ***How does a bacterial population grow?***

- a. c. Conclusion: the number of cells will double with each new generation (exponential growth).

- d. Exponential growth equation

$$N_2 = N_1 2^n \quad \text{equation (1)}$$

$N_2$  = the number of bacterial cells at time  $t_2$

$N_1$  = the number of bacterial cells at time  $t_1$

$n$  = the number of doublings

- e. The logarithmic form of the equation:

$$\log N_2 = \log N_1 + n \log 2 \quad \text{equation (2)}$$

- f. Definition of doubling time:

$$g = (t_2 - t_1) / n \quad \text{equation (3)}$$

- g. Substitute equation (3) into equation (2):

$$\log N_2 = \log N_1 + \log 2 (t_2 - t_1) / g \quad \text{equation (4)}$$

- h. Substitute  $\log 2 = 0.30$ :

$$\log N_2 = \log N_1 + 0.30 (t_2 - t_1) / g \quad \text{equation (5)}$$

- i. Usually  $N_2$ ,  $N_1$ , and  $(t_2 - t_1)$  are measured experimentally and we solve equation (5) for  $g$ . However, we can solve the equation for any unknown if all of the other quantities are given.

- j. Example: for a culture of *E. coli*: let  $g = 20$  minutes,  $N_1 = 1$  cell, and  $(t_2 - t_1) = 2$  days (2,880 minutes). What is  $N_2$ ?

$$(\log 1 = 0)$$

$$\log N_2 = (\log 1) + 0.30 (2,880 \text{ minutes}) / (20 \text{ minutes}) = 43.2$$

$$N_2 = 1.6 \times 10^{43} \text{ cells}$$

If each cell weighs  $1 \times 10^{-12}$  grams:

$$(1.6 \times 10^{43} \text{ cells}) (1 \times 10^{-12} \text{ g / cell}) = 1.6 \times 10^{31} \text{ g}$$

= approximately 2,700 times the total mass of the Earth ( $6 \times 10^{27}$  grams)